

# Pd(II)-Catalyzed Enantioselective C(sp<sup>3</sup>)-H Borylation

Jian He,<sup>b</sup> Qian Shao,<sup>†</sup> Qingfeng Wu,<sup>†</sup> and Jin-Quan Yu\*<sup>b</sup>

Department of Chemistry, The Scripps Research Institute, 10550 North Torrey Pines Road, La Jolla, California 92037, United States

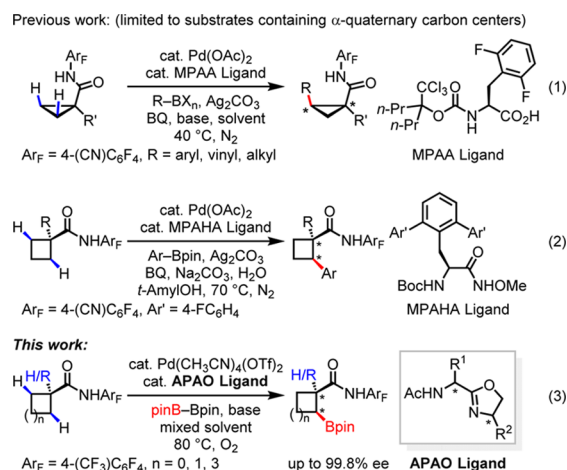
**S** Supporting Information

**ABSTRACT:** Pd(II)-catalyzed enantioselective borylation of C(sp<sup>3</sup>)-H bonds has been realized for the first time using chiral acetyl-protected aminomethyl oxazoline ligands. This reaction is compatible with carbocyclic amides containing  $\alpha$ -tertiary as well as  $\alpha$ -quaternary carbon centers. The chiral  $\beta$ -borylated amides are useful synthons for the synthesis of chiral  $\beta$ -hydroxylated,  $\beta$ -fluorinated, and  $\beta$ -arylated carboxylic acids.

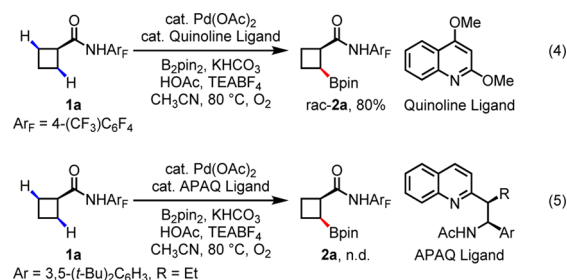
Differentiation of prochiral C-H bonds through metal insertion has recently emerged as a promising approach in asymmetric catalysis.<sup>1,2</sup> A number of Pd(0)-catalyzed enantioselective intramolecular C(sp<sup>3</sup>)-H arylation and alkylation reactions have been realized by the use of chiral *N*-heterocyclic carbene<sup>3</sup> or phosphine ligands.<sup>4,5</sup> The feasibility of Pd(II)-catalyzed enantioselective intermolecular C(sp<sup>3</sup>)-H activation was initially demonstrated using mono-*N*-protected amino acid (MPAA) ligands.<sup>6</sup> Major advances have been recently made by using a weakly coordinating monodentate substrate and a chiral bidentate ligand. The design of chiral bidentate acetyl-protected aminoethyl quinoline (APAQ) and mono-*N*-protected aminoethyl oxazoline (MPAO) ligands for asymmetric induction have led to the development of enantioselective intermolecular arylation of methylene C(sp<sup>3</sup>)-H bonds and *gem*-dimethyl C(sp<sup>3</sup>)-H bonds, respectively.<sup>7,8</sup>

The development of these enantioselective C(sp<sup>3</sup>)-H activation reactions involving Pd(II)/Pd(IV) redox catalysis calls into question whether enantioselective C(sp<sup>3</sup>)-H activation reactions with nucleophiles through Pd(II)/Pd(0) redox catalysis are compatible with the APAQ and MPAO ligands. Although both catalytic reactions proceed through the same asymmetric C-H insertion intermediates, the transmetalation and reductive elimination steps in Pd(II)/Pd(0) catalytic cycles may require different ligand scaffolds. Enantioselective C(sp<sup>3</sup>)-H cross-coupling reactions via Pd(II)/Pd(0) catalysis have been developed using MPAA and mono-*N*-protected  $\alpha$ -amino-*O*-methylhydroxamic acid (MPAHA) ligands, albeit with significant limitations (Scheme 1, eqs 1 and 2).<sup>6,9</sup> Notably, enantioselective C(sp<sup>3</sup>)-H borylation has not been developed to date.<sup>10-12</sup> Herein, we report the first example of Pd(II)-catalyzed enantioselective  $\beta$ -borylation of carboxylic acid derived amides with bis(pinacolato)diboron using acetyl-protected aminomethyl oxazoline (APAO) ligands (Scheme 1, eq 3). A range of cyclic amides, including cyclopropanes, cyclobutanes, and cyclohexanes, can be successfully borylated with high levels of enantioselectivity. This reaction is compatible with amide substrates containing  $\alpha$ -tertiary as well as  $\alpha$ -quaternary carbon

## Scheme 1. Enantioselective C(sp<sup>3</sup>)-H Activation via Pd(II)/Pd(0) Catalysis



## Scheme 2. Palladium-Catalyzed C(sp<sup>3</sup>)-H Borylation Using Quinoline-Based Ligands

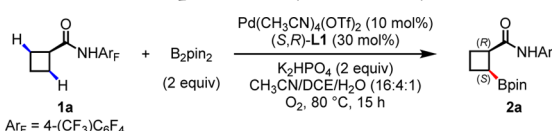


centers. Notably, O<sub>2</sub> is used as the sole oxidant for this Pd(II)/Pd(0) catalysis.

We have previously developed C(sp<sup>3</sup>)-H borylation promoted by a monodentate quinoline ligand (Scheme 2, eq 4).<sup>12f</sup> Since the recently developed chiral APAQ ligands for enantioselective  $\beta$ -C-H arylation also contain a quinoline moiety,<sup>7</sup> we initiated our investigation on the borylation of **1a** using this type of ligands (Scheme 2, eq 5). Although these bidentate quinoline ligands are known to promote C(sp<sup>3</sup>)-H cleavage of amide substrates, they failed to provide any desired borylated products under our previously established conditions. It appears that this ligand scaffold is not compatible with the transmetalation or the C(sp<sup>3</sup>)-B reductive elimination step.

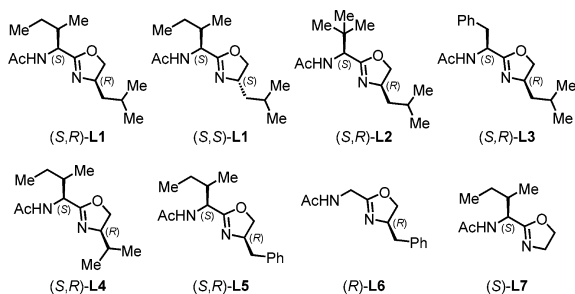
Received: January 4, 2017

Published: February 17, 2017

**Table 1. Effect of Reaction Parameters in Pd(II)-Catalyzed Enantioselective C(sp<sup>3</sup>)-H Borylation of Cyclic Amide **1a**<sup>a,b</sup>**


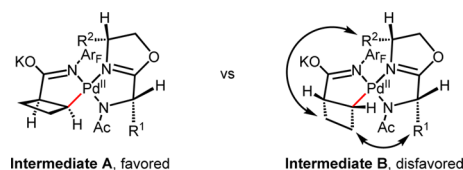
entry	variation from standard conditions	ee (%)	yield (%)
1	none	95.6	82
2	no Pd(CH <sub>3</sub> CN) <sub>4</sub> (OTf) <sub>2</sub>	–	n.d.
3	no K <sub>2</sub> HPO <sub>4</sub>	–	n.d.
4	no (S,R)-L1	0	21
5	(S,R)-L1 (20 mol %)	93.4	85
6	(S,S)-L1, instead of (S,R)-L1	50.4	24
7	(S,R)-L2, instead of (S,R)-L1	95.0	82
8	(S,R)-L3, instead of (S,R)-L1	93.6	72
g	(S,R)-L4, instead of (S,R)-L1	92.6	71
10	(S,R)-L5, instead of (S,R)-L1	92.2	76
11	(R)-L6, instead of (S,R)-L1	51.0	13
12	(S)-L7, instead of (S,R)-L1	78.8	57
13 <sup>c</sup>	Pd(CH <sub>3</sub> CN) <sub>4</sub> (OTf) <sub>2</sub> (5 mol %)	89.0	67
14	Pd(OAc) <sub>2</sub> , instead of Pd(CH <sub>3</sub> CN) <sub>4</sub> (OTf) <sub>2</sub>	78.4	63
15	KHCO <sub>3</sub> , instead of K <sub>2</sub> HPO <sub>4</sub>	89.4	58
16	CH <sub>3</sub> CN only	93.4	64
17	DCE only	72.4	19
18	CH <sub>3</sub> CN/DCE (4:1)	95.6	65
19	60 °C, instead of 80 °C	95.4	38
20	under air (capped vial)	94.8	61

<sup>a</sup>Reaction conditions: substrate **1a**, B<sub>2</sub>pin<sub>2</sub> (2.0 equiv), Pd(CH<sub>3</sub>CN)<sub>4</sub>(OTf)<sub>2</sub> (10 mol %), APAO ligand (30 mol %), K<sub>2</sub>HPO<sub>4</sub> (2.0 equiv), CH<sub>3</sub>CN/DCE/H<sub>2</sub>O (16:4:1), O<sub>2</sub>, 80 °C, 15 h. <sup>b</sup>The yield was determined by <sup>1</sup>H NMR of the crude product using CH<sub>2</sub>Br<sub>2</sub> as the internal standard. The ee values were determined by HPLC analysis on a chiral stationary phase. <sup>c</sup>(S,R)-L1 (15 mol %) was used.



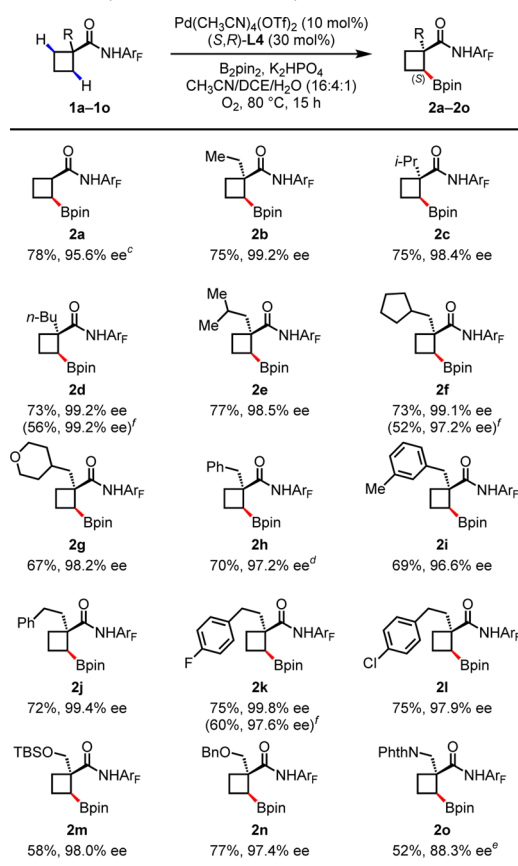
Considering the previously observed significant steric effect of ligands on C(sp<sup>3</sup>)-H borylation,<sup>12f</sup> we replaced the quinoline motif with a variety of heterocycles. We found acetyl-protected aminomethyl oxazolines as the only effective ligands for this unprecedented enantioselective C(sp<sup>3</sup>)-H borylation reaction. The enantioenriched borylated product **2a** was obtained in 82% <sup>1</sup>H NMR yield and 95.6% ee when reacting **1a** with bis(pinacolato)diboron in the presence of Pd(CH<sub>3</sub>CN)<sub>4</sub>(OTf)<sub>2</sub> (10 mol %), bidentate oxazoline ligand (S,R)-L1 (30 mol %), and K<sub>2</sub>HPO<sub>4</sub> in the mixed solvent (CH<sub>3</sub>CN/DCE/H<sub>2</sub>O) at 80 °C under O<sub>2</sub> for 15 h (Table 1, entry 1). Control experiments revealed that the Pd catalyst and base were crucial for the C(sp<sup>3</sup>)-H borylation to proceed (entries 2 and 3). Low reaction conversion was observed in the absence of ligands (entry 4). Decreasing the loading of (S,R)-L1 to 20 mol % slightly increased the yield of **2a**, but the ee dropped to 93.4% (entry 5). The use of (S,S)-L1, the other diastereomer of the optimal ligand, drastically decreased the yield and ee to 24% and 50.4%, respectively (entry

6). Changing the substituents on both the side chain and the oxazoline moiety of (S,R)-APAO ligands did not have a significant impact on reactivity and enantioselectivity (entries 7–10). Yet, the chiral oxazoline ligands lacking a stereocenter on either the side chain or the oxazoline moiety gave poor enantioselectivity (entries 11 and 12), which speaks to the importance of both chiral centers on the APAO ligand backbones for asymmetric induction. The observed significant impact of both chiral centers on the enantioselectivity can be explained by the potential structures of the C–H insertion intermediates shown in Figure 1. When (S,R)-

**Figure 1.** Proposed asymmetric induction model in enantioselective C(sp<sup>3</sup>)-H borylation of cyclobutanes.

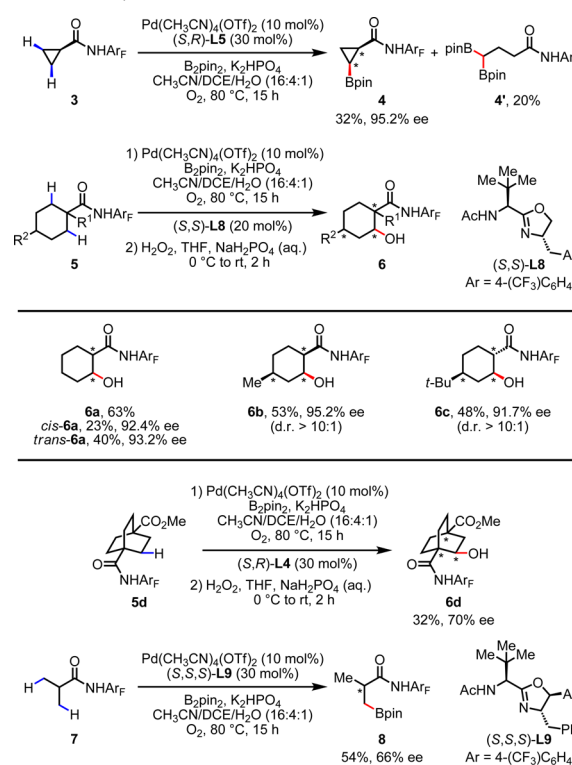
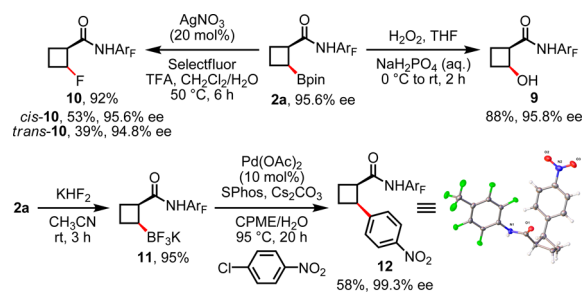
oxazoline ligands are employed in C(sp<sup>3</sup>)-H borylation, the substituents on both chiral centers of the ligand backbone exert a synergistic effect in asymmetric induction through the steric repulsion with the cyclobutane ring, favoring five-membered palladacycle **Intermediate A** over **Intermediate B**. Reducing the amount of Pd(CH<sub>3</sub>CN)<sub>4</sub>(OTf)<sub>2</sub> to 5 mol % gave **2a** in 67% yield and 89.0% ee (entry 13). The fine-tuning of palladium sources, bases, and solvents was also essential for achieving high levels of enantioselectivity in this C(sp<sup>3</sup>)-H borylation (entries 14–18). A similar ee value was obtained when conducting the reaction at 60 °C, but the yield was significantly lower (entry 19). The reaction could also be carried out in air, affording the desired product in 61% yield and 94.8% ee (entry 20).

With the optimal reaction conditions in hand, we then evaluated the scope of cyclobutanecarboxylic acid derived amides toward enantioselective C(sp<sup>3</sup>)-H borylation (Table 2). While (S,R)-L1 was superior in the borylation of amide substrate **1a** bearing an  $\alpha$ -H atom, (S,R)-L4 containing an isopropyl group on the oxazoline moiety gave better enantioselectivity for substrates containing  $\alpha$ -quaternary C-centers in general. Simple  $\alpha$ -alkyl substituted amides could be borylated in good yields with excellent levels of enantioselectivity, ranging from 98.4% to 99.2% ee (**2b–2f**). A variety of aromatic rings at the  $\beta$ - or  $\gamma$ -positions of the exocyclic alkyl side chain were well tolerated (**2h–2i**). C(sp<sup>2</sup>)-H borylation was not detected in the functionalization of these substrates. (S,R)-L5 gave the highest enantioselectivity in the borylation of amide **1h** containing a simple benzyl group at the  $\alpha$ -position. Substrates with potentially coordinating heteroatoms such as oxygen (**1g**, **1m**, and **1n**) and nitrogen (**1o**) on the  $\alpha$ -substituents were also compatible with this enantioselective C(sp<sup>3</sup>)-H borylation. Even though water had to be excluded from the solvent mixture to prevent hydrolysis of the phthalimido group, C(sp<sup>3</sup>)-H borylation of  $\beta$ -amino acid substrate **1o** still provided the desired product **2o** in a moderate yield. The absolute configuration of the borylated compounds was confirmed by X-ray crystallographic analysis after an oxidation sequence (Supporting Information). To probe the efficiency of this catalytic system, we also carried out the enantioselective C(sp<sup>3</sup>)-H borylation in the presence of 5 mol % of Pd(CH<sub>3</sub>CN)<sub>4</sub>(OTf)<sub>2</sub> and 15 mol % of (S,R)-L4, which provided the desired products (**2d**, **2f**, and **2k**) in 52–60% yield and 97.2–99.2% ee.

**Table 2. Substrate Scope for Enantioselective C(sp<sup>3</sup>)-H Borylation of Cyclobutanecarboxylic Amides<sup>a,b</sup>**


<sup>a</sup>Reaction conditions: substrate **1a–1o**, B<sub>2</sub>pin<sub>2</sub> (2.0 equiv), Pd(CH<sub>3</sub>CN)<sub>4</sub>(OTf)<sub>2</sub> (10 mol %), (S,R)-L4 (30 mol %), K<sub>2</sub>HPO<sub>4</sub> (2.0 equiv), CH<sub>3</sub>CN/DCE/H<sub>2</sub>O (16:4:1), O<sub>2</sub>, 80 °C, 15 h. <sup>b</sup>Isolated yields. The ee values were determined by HPLC analysis on a chiral stationary phase. <sup>c</sup>(S,R)-L1 (30 mol %) was used. <sup>d</sup>(S,R)-L5 (30 mol %) was used. <sup>e</sup>CH<sub>3</sub>CN/DCE (4:1) was used as the mixed solvent. <sup>f</sup>Pd(CH<sub>3</sub>CN)<sub>4</sub>(OTf)<sub>2</sub> (5 mol %) and (S,R)-L4 (15 mol %) were used.

Under similar reaction conditions using (S,R)-L5 as the optimal ligand, cyclopropanecarboxylic amide **3** was borylated in 32% yield and 95.2% ee, while an unexpected diborylated product **4'** was also obtained in 20% isolated yield (Scheme 3). Pd(CH<sub>3</sub>CN)<sub>4</sub>(OTf)<sub>2</sub> was found to promote the side product formation, but the detailed reaction mechanism remains to be exploited. In the borylation of cyclohexane and *gem*-dimethyl substrates (**5a–5c** and **7**), (S,S)-APAO ligands were substantially more effective than their (S,R)-diastereomers, which is consistent with the ligand effect in the enantioselective C(sp<sup>3</sup>)-H arylation of amide **7** with aryl iodides.<sup>8</sup> The use of (S,S)-L8 (20 mol %) to activate cyclohexanecarboxylic amide **5a** gave the chiral borylated products as a *cis*- and *trans*-mixture which were then oxidized to provide *cis*-**6a** and *trans*-**6a** in 92.4% and 93.2% ee, respectively. The borylation/oxidation of both *cis*- and *trans*-4-substituted cyclohexane substrates furnished the desired products (**6b** and **6c**) with high levels of diastereoselectivity (d.r. > 10:1) and enantioselectivity (95.2% and 91.7% ee, respectively). The reaction conditions were also applied to the asymmetric C(sp<sup>3</sup>)-H functionalization of the bicyclo[2.2.2]octane (**5d**), giving the enantioenriched hydroxylated product (**6d**) in 70% ee. With the aid of (S,S,S)-L9, desymmetrization of the isopropyl

**Scheme 3. Enantioselective C(sp<sup>3</sup>)-H Borylation of Other Cyclic and Acyclic Amides**

**Scheme 4. Synthetic Applications of Enantioselective C(sp<sup>3</sup>)-H Borylation**


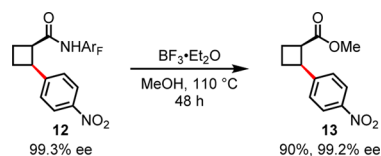
group in isobutyric acid derivative **7** afforded **8** in moderate yield (54%) and enantioselectivity (66% ee).

To further demonstrate the synthetic utility of this enantioselective C(sp<sup>3</sup>)-H borylation reaction,<sup>13</sup> we subjected the chiral cyclobutylboronate ester **2a** to various reaction conditions, constructing a range of  $\beta$ -chiral centers containing C–heteroatom or C–C bonds (Scheme 4). After reacting **2a** with hydrogen peroxide in a THF and NaH<sub>2</sub>PO<sub>4</sub> aqueous solution, the desired  $\beta$ -hydroxylated product **9** could be obtained in 88% yield and 95.8% ee. The combination of AgNO<sub>3</sub> and Selectfluor in CH<sub>2</sub>Cl<sub>2</sub>/H<sub>2</sub>O (1:1) converted **2a** into chiral  $\beta$ -fluorinated cyclobutanecarboxylic acid derivatives (*cis*-**10** and *trans*-**10**) in a total yield of 92% through a radical pathway.<sup>14</sup> High ee values were maintained in both diastereomers. Treatment of the borylated product **2a** with KHF<sub>2</sub> in acetonitrile led to the formation of trifluoroborate salt **11** in 95% yield after recrystallization. The Suzuki–Miyaura cross-coupling between 1-chloro-4-nitrobenzene and **11** provided the  $\beta$ -arylated product **12** in 99.3% ee with retention of stereochemistry.<sup>15</sup> The absolute configuration of **12** was also confirmed by X-ray crystallographic



analysis. Removal of the amide auxiliary by  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  in methanol gave the corresponding ester **13** in 90% yield without loss of enantiomeric purity (Scheme 5).<sup>16</sup>

### Scheme 5. Removal of the Amide Auxiliary



In summary, we have developed enantioselective  $\text{C}(\text{sp}^3)\text{-H}$  borylation of weakly coordinating carboxylic amides via a  $\text{Pd}(\text{II})/\text{Pd}(0)$  catalytic cycle. Pivotal to the success of this asymmetric catalysis was the use of chiral bidentate APAO ligands. This reaction is compatible with carbocyclic substrates containing  $\alpha$ -tertiary as well as  $\alpha$ -quaternary carbon centers. The borylated products can be converted into various chiral  $\beta$ -hydroxylated,  $\beta$ -fluorinated, and  $\beta$ -arylated carboxylic acids. Development of more effective chiral bidentate ligands to enable highly enantioselective borylation of *gem*-dimethyl  $\text{C}(\text{sp}^3)\text{-H}$  bonds is currently underway in our laboratory.

## ■ ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/jacs.6b13389.

Experimental procedures and spectral data (PDF)  
Crystallographic data (CIF, CIF, CIF)

## ■ AUTHOR INFORMATION

### Corresponding Author

\*yu200@scripps.edu

### ORCID

Jian He: 0000-0002-3388-3239

Jin-Quan Yu: 0000-0003-3560-5774

### Author Contributions

†Q.S. and Q.W. contributed equally.

### Notes

The authors declare no competing financial interest.

## ■ ACKNOWLEDGMENTS

We gratefully acknowledge The Scripps Research Institute and the NIH (NIGMS, 2R01GM084019) for financial support.

## ■ REFERENCES

- (1) (a) Daugulis, O.; Do, H.-Q.; Shabashov, D. *Acc. Chem. Res.* **2009**, *42*, 1074. (b) Wasa, M.; Engle, K. M.; Yu, J.-Q. *Isr. J. Chem.* **2010**, *50*, 605. (c) Lyons, T. W.; Sanford, M. S. *Chem. Rev.* **2010**, *110*, 1147. (d) Baudoin, O. *Chem. Soc. Rev.* **2011**, *40*, 4902. (e) Daugulis, O.; Roane, J.; Tran, L. D. *Acc. Chem. Res.* **2015**, *48*, 1053. (f) Pedroni, J.; Cramer, N. *Chem. Commun.* **2015**, *51*, 17647. (g) He, G.; Wang, B.; Nack, W. A.; Chen, G. *Acc. Chem. Res.* **2016**, *49*, 635. (h) He, J.; Wasa, M.; Chan, K. S. L.; Shao, Q.; Yu, J.-Q. *Chem. Rev.* **2017**, ASAP, DOI: 10.1021/acs.chemrev.6b00622.
- (2) (a) Shi, B.-F.; Mangel, N.; Zhang, Y.-H.; Yu, J.-Q. *Angew. Chem., Int. Ed.* **2008**, *47*, 4882. (b) Gao, D.-W.; Shi, Y.-C.; Gu, Q.; Zhao, Z.-L.; You, S.-L. *J. Am. Chem. Soc.* **2013**, *135*, 86. (c) Yan, S.-B.; Zhang, S.; Duan, W.-L. *Org. Lett.* **2015**, *17*, 2458. (d) Wang, H.; Tong, H.-R.; He, G.; Chen, G. *Angew. Chem., Int. Ed.* **2016**, *55*, 15387.
- (3) (a) Nakanishi, M.; Katayev, D.; Besnard, C.; Kündig, E. P. *Angew. Chem., Int. Ed.* **2011**, *50*, 7438. (b) Katayev, D.; Nakanishi, M.; Buergi, T.;

Kündig, E. P. *Chem. Sci.* **2012**, *3*, 1422. (c) Larionov, E.; Nakanishi, M.; Katayev, D.; Besnard, C.; Kündig, E. P. *Chem. Sci.* **2013**, *4*, 1995. (d) Katayev, D.; Larionov, E.; Nakanishi, M.; Besnard, C.; Kündig, E. P. *Chem. - Eur. J.* **2014**, *20*, 15021.

(4) (a) Anas, S.; Cordi, A.; Kagan, H. B. *Chem. Commun.* **2011**, *47*, 11483. (b) Saget, T.; Lemouzy, S. J.; Cramer, N. *Angew. Chem., Int. Ed.* **2012**, *51*, 2238. (c) Martin, N.; Pierre, C.; Davi, M.; Jazzar, R.; Baudoin, O. *Chem. - Eur. J.* **2012**, *18*, 4480.

(5) (a) Saget, T.; Cramer, N. *Angew. Chem., Int. Ed.* **2012**, *51*, 12842. (b) Pedroni, J.; Boghi, M.; Saget, T.; Cramer, N. *Angew. Chem., Int. Ed.* **2014**, *53*, 9064. (c) Pedroni, J.; Saget, T.; Donets, P. A.; Cramer, N. *Chem. Sci.* **2015**, *6*, 5164. (d) Pedroni, J.; Cramer, N. *Angew. Chem., Int. Ed.* **2015**, *54*, 11826.

(6) Wasa, M.; Engle, K. M.; Lin, D. W.; Yoo, E. J.; Yu, J.-Q. *J. Am. Chem. Soc.* **2011**, *133*, 19598.

(7) Chen, G.; Gong, W.; Zhuang, Z.; Andrä, M. S.; Chen, Y.-Q.; Hong, X.; Yang, Y.-F.; Liu, T.; Houk, K. N.; Yu, J.-Q. *Science* **2016**, *353*, 1023.

(8) Wu, Q.-F.; Shen, P.-X.; He, J.; Wang, X.-B.; Zhang, F.; Shao, Q.; Zhu, R.-Y.; Mapelli, C.; Qiao, J. X.; Poss, M. A.; Yu, J.-Q. *Science* **2017**, *355*, 499.

(9) Xiao, K.-J.; Lin, D. W.; Miura, M.; Zhu, R.-Y.; Gong, W.; Wasa, M.; Yu, J.-Q. *J. Am. Chem. Soc.* **2014**, *136*, 8138.

(10) (a) Mkhali, I. A. I.; Barnard, J. H.; Marder, T. B.; Murphy, J. M.; Hartwig, J. F. *Chem. Rev.* **2010**, *110*, 890. (b) Cho, J.-Y.; Tse, M. K.; Holmes, D.; Maleczka, R. E., Jr.; Smith, M. R. *Science* **2002**, *295*, 305.

(11) Representative examples of transition-metal-catalyzed non-directed  $\text{C}(\text{sp}^3)\text{-H}$  borylation: (a) Chen, H.; Hartwig, J. F. *Angew. Chem., Int. Ed.* **1999**, *38*, 3391. (b) Chen, H.; Schlecht, S.; Semple, T. C.; Hartwig, J. F. *Science* **2000**, *287*, 1995. (c) Shimada, S.; Batsanov, A. S.; Howard, J. A. K.; Marder, T. B. *Angew. Chem., Int. Ed.* **2001**, *40*, 2168. (d) Murphy, J. M.; Lawrence, J. D.; Kawamura, K.; Incarvito, C.; Hartwig, J. F. *J. Am. Chem. Soc.* **2006**, *128*, 13684. (e) Liskey, C. W.; Hartwig, J. F. *J. Am. Chem. Soc.* **2012**, *134*, 12422. (f) Palmer, W. N.; Obligation, J. V.; Pappas, I.; Chirik, P. J. *J. Am. Chem. Soc.* **2016**, *138*, 766.

(12) For representative examples of transition-metal-catalyzed directed  $\text{C}(\text{sp}^3)\text{-H}$  borylation, see: (a) Kawamorita, S.; Miyazaki, T.; Iwai, T.; Ohmiya, H.; Sawamura, M. *J. Am. Chem. Soc.* **2012**, *134*, 12924. (b) Kawamorita, S.; Murakami, R.; Iwai, T.; Sawamura, M. *J. Am. Chem. Soc.* **2013**, *135*, 2947. (c) Cho, S. H.; Hartwig, J. F. *J. Am. Chem. Soc.* **2013**, *135*, 8157. (d) Zhang, L.-S.; Chen, G.; Wang, X.; Guo, Q.-Y.; Zhang, X.-S.; Pan, F.; Chen, K.; Shi, Z.-J. *Angew. Chem., Int. Ed.* **2014**, *53*, 3899. (e) Miyamura, S.; Araki, M.; Suzuki, T.; Yamaguchi, J.; Itami, K. *Angew. Chem., Int. Ed.* **2015**, *54*, 846. (f) He, J.; Jiang, H.; Takise, R.; Zhu, R.-Y.; Chen, G.; Dai, H.-X.; Dhar, T. G. M.; Shi, J.; Zhang, H.; Cheng, P. T. W.; Yu, J.-Q. *Angew. Chem., Int. Ed.* **2016**, *55*, 785. (g) Larsen, M. A.; Cho, S. H.; Hartwig, J. F. *J. Am. Chem. Soc.* **2016**, *138*, 762.

(13) (a) Hall, D. G., Ed. *Boronic Acids: Preparation and Applications in Organic Synthesis and Medicine*; Wiley-VCH: Weinheim, Germany, 2005. (b) Miyaura, N.; Suzuki, A. *Chem. Rev.* **1995**, *95*, 2457. (c) Hayashi, T.; Yamasaki, K. *Chem. Rev.* **2003**, *103*, 2829. (d) Jana, R.; Pathak, T. P.; Sigman, M. S. *Chem. Rev.* **2011**, *111*, 1417. (e) Kirchhoff, J. H.; Netherton, M. R.; Hills, I. D.; Fu, G. C. *J. Am. Chem. Soc.* **2002**, *124*, 13662. (f) Saito, B.; Fu, G. C. *J. Am. Chem. Soc.* **2007**, *129*, 9602. (g) Mlynarski, S. N.; Schuster, C. H.; Morken, J. P. *Nature* **2013**, *505*, 386. (h) Tellis, J. C.; Primer, D. N.; Molander, G. A. *Science* **2014**, *345*, 433. (i) Fang, L.; Yan, L.; Haeflner, F.; Morken, J. P. *J. Am. Chem. Soc.* **2016**, *138*, 2508.

(14) (a) Li, Z.; Wang, Z.; Zhu, L.; Tan, X.; Li, C. *J. Am. Chem. Soc.* **2014**, *136*, 16439. (b) Ramírez, J.; Fernández, E. *Tetrahedron Lett.* **2007**, *48*, 3841.

(15) (a) Ohmura, T.; Awano, T.; Sugimoto, M. *J. Am. Chem. Soc.* **2010**, *132*, 13191. (b) Sandrock, D. L.; Jean-Gérard, L.; Chen, C.-Y.; Dreher, S. D.; Molander, G. A. *J. Am. Chem. Soc.* **2010**, *132*, 17108. (c) Lee, J. C. H.; McDonald, R.; Hall, D. G. *Nat. Chem.* **2011**, *3*, 894. (d) Awano, T.; Ohmura, T.; Sugimoto, M. *J. Am. Chem. Soc.* **2011**, *133*, 20738. (e) Li, L.; Zhao, S.; Joshi-Pangu, A.; Diane, M.; Biscoe, M. R. *J. Am. Chem. Soc.* **2014**, *136*, 14027.

(16) He, J.; Li, S.; Deng, Y.; Fu, H.; Laforteza, B. N.; Spangler, J. E.; Homs, A.; Yu, J.-Q. *Science* **2014**, *343*, 1216.